

Multiple Pulse Proton Magnetic Resonance Study of KHCO_3 and KHSO_4

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(Z. Naturforsch. **31a**, 1173—1180 [1976]; received August 3, 1976)

Single crystals of KHCO_3 and KHSO_4 have been investigated by solid state high resolution proton magnetic resonance techniques. These compounds may be considered as model systems for compounds that contain abundant quadrupolar nuclei (^{39}K) with small magnetogyric ratio in addition to the protons. Despite considerable line broadening resulting from ^1H - ^{39}K dipole-dipole coupling, all four possible proton shielding tensors in KHSO_4 , and an averaged shielding tensor for the two different proton sites in KHCO_3 could be measured accurately. The results are analysed to determine the degree of ionization of the protons in the three types of hydrogen bonds occurring in KHCO_3 and KHSO_4 .

1. Introduction

Multiple pulse techniques for high resolution NMR in solids have now been applied successfully for measuring proton shielding tensors in a variety of solids^{1, 2}. The simplest contained no other nuclear species with magnetic moments in addition to the protons so suppression of the ^1H — ^1H dipole-dipole interactions with any of the standard multiple pulse sequences¹ did lead to “high-resolution” spectra. Others contained, in addition to the protons, further abundant spin-1/2-nuclei, e.g. ^{19}F or ^{31}P . Simultaneous homo-hetero-decoupling has been demonstrated to work in these cases^{3–5}. On the other hand, it does not seem feasible *in general* to decouple quadrupolar nuclei from protons in solids. Exceptions are ^2H and possibly a few other nuclei with small nuclear quadrupole moment eQ ⁶. For some nuclei with very large eQ , or more precisely, with very large quadrupole coupling constants e^2qQ , heterodecoupling may not be needed for achieving “high resolution” in the proton spectrum because of the phenomenon of “selfdecoupling”⁷. Here we describe the investigation of single crystals of KHCO_3 and KHSO_4 where the protons are dipole coupled to the abundant quadrupolar ^{39}K -nuclei. KHCO_3 and KHSO_4 may be considered to be typical representatives of the class of compounds that contain abundant $I > 1/2$ nuclei with small mag-

netic moments — the magnetic moment of ^{39}K is only 0.4 nuclear magnetons — and for which no effective selfdecoupling may be expected. Our results show that accurate measurements of proton shielding tensors can be accomplished in such compounds even in the presence of heteronuclear dipolar line broadening.

All the protons in KHCO_3 and KHSO_4 are involved in hydrogen bonds. In KHSO_4 there are two different types of hydrogen bonds (see below). Any study of a hydrogen bond by its shielding tensor beyond just establishing numbers requires either an empirical or a theoretical basis. Gierke and Flygare⁸ have put forth a simple theory for shielding tensors whose application to protons forming hydrogen bonds has revealed, among other things, the physical origin of the dominant contributor to the proton shielding anisotropy: it is the Larmor-precession induced by the applied field B_0 of the electrons of the hydrogen bonded atoms¹. This has been confirmed recently by much more sophisticated ab-initio calculations of some model hydrogen bonds carried out by Ditchfield et al.⁹. This adds to our confidence to use the GF-theory for extracting information from our shielding data on the degree of ionization of the hydrogens in the three types of hydrogen bonds studied in this work.

2. Crystal Structures, Number of NMR Lines Expected from Crystal Symmetry

(a) KHCO_3

Crystals of KHCO_3 are monoclinic, space group $\text{P}2_1/c$, $a = 15.17 \text{ \AA}$, $b = 5.63 \text{ \AA}$, $c = 3.17 \text{ \AA}$, $\beta = 104.6^\circ$ ¹⁰, with $Z = 4$ molecules in the unit cell. The

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HCO_3^- -anions form dimers, linked by two adjacent hydrogen bonds. The two protons of each dimer are related by a center of symmetry (see Fig. 1) and are, therefore, magnetically equivalent. As there are two orientationally inequivalent dimers, the solid

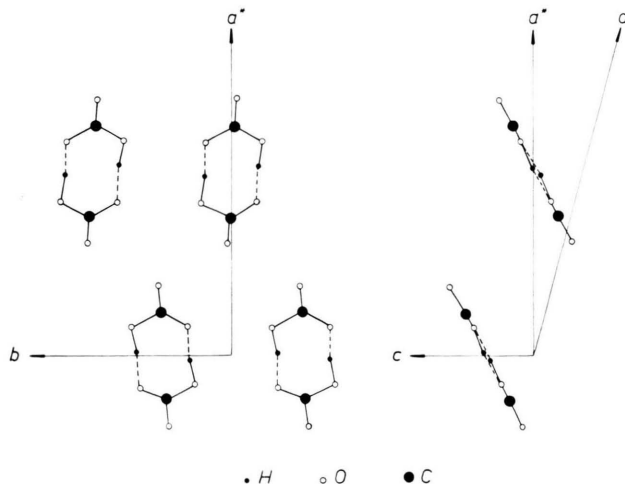


Fig. 1. Arrangement of $[\text{HCO}_3^-]_2$ -dimers in the crystal lattice of KHCO_3 ; the two halves of the dimer are related by inversion centers.

state high resolution spectrum is expected to consist of two lines in general. The two types of dimers are related by the monoclinic glide plane (the a^* - c -plane) so all protons in KHCO_3 have the same set of principal shielding components and one line only is expected in the proton spectrum for the special case when B_0 lies in the glide plane. As regards the principal shielding directions we expected from the results of previous solid state PMR studies of hydrogen bonds (for a review see Ref. ¹) that the most shielded directions would be approximately parallel to the directions of the hydrogen bonds and that the least shielded directions would be close to the normals of the quasi-planes formed by the dimers. As Fig. 1 shows, all these directions lie for both types of dimers not far from the monoclinic glide plane. A rotation of the crystal about the b -axis, i. e. for B_0 moving in the a^* - c plane, should therefore directly reveal approximate values for two of the three principal shielding components. On the other hand, a rotation about the c -axis is most suited to detect, by a splitting of the proton resonance, the expected difference of the orientations of the most shielded proton directions for the two types of dimers. For these reasons we choose the b and c axes as rotation

axes for two of our samples. In order to have an orthogonal set of three axes we choose a^* as the rotation axis for a third sample. These samples will be referred to as a^* , b and c crystals.

(b) KHSO_4

Crystals of KHSO_4 are orthorhombic, space group $Pbca$, $a = 8.40 \text{ \AA}$, $b = 9.79 \text{ \AA}$, $c = 18.93 \text{ \AA}$; $Z = 16$ ¹¹. The system of hydrogen bonds is particularly interesting in KHSO_4 : the crystal may be considered as consisting of two independent but mutually interpenetrating sublattices, one of which is formed by $(\text{HSO}_4^-)_2$ dimers held together by two adjacent hydrogen bonds similar as in KHCO_3 , whereas the other consists of infinite chains, also linked by hydrogen bonds (see Figure 2). Of both the dimers

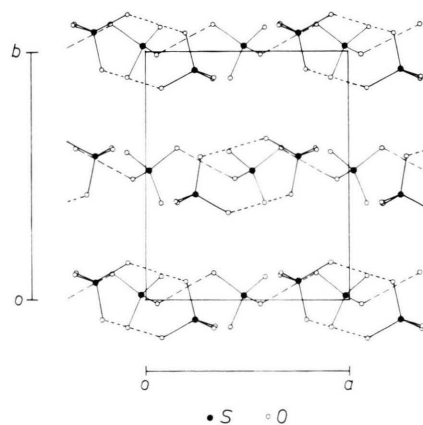


Fig. 2. Arrangement of $[\text{HSO}_4^-]_n$ -chains and $[\text{HSO}_4^-]_2$ -dimers in alternating layers in the crystal lattice of KHSO_4 . The protons are not shown, actually, their positions have not been determined by X-rays. Hydrogen bonded oxygens are connected by dashed lines.

and the chains there are two orientationally (magnetically) inequivalent species, but the protons of all the dimers and of all the chains, respectively, are crystallographically equivalent, so we expect four different proton shielding tensors in KHSO_4 , but only two different sets of principal components.

As all the hydrogen bonds are lying approximately in the a - b -plane, it is but natural to use the a , b , and c axes as rotation axes for the NMR work. For analogous reasons as outlined above for KHCO_3 rotation of the KHSO_4 crystal about the c -axis (B_0 moving in the a - b -plane) should and does give the most valuable information.

3. Experimental

3.1. Preparation of Samples

KHCO_3 : Three samples were cut from one large single crystal of KHCO_3 obtained from Prof. Hausühl, University of Cologne. They were preoriented using an optical goniometer, shaped into cylinders (4 mm diameter, 6 mm length), precisely oriented by means of X-ray precession photographs and glued to glass rods carrying discs with angular scales such that their a^* , b , and c -axes, respectively, were parallel to the rod axes. The relatively large linewidths in the "high resolution" solid state proton spectra did not warrant the effort to shape spheres for eliminating bulk susceptibility effects¹².

KHSO_4 : A single crystal cube (length of edge = 4 mm) of KHSO_4 was purchased from Minhorst, Kristallchemie GmbH, D-5431 Meudt, FRG. It was cut such that its edges were parallel to the crystallographic a , b , and c -axes. It was glued with one of its faces to the end of a glass rod as were the KHCO_3 crystals. After completing the NMR data acquisition for this rotation axis, it was removed and reglued with a different face. The procedure was then repeated for the third axis. The orientation was checked each time as described for KHCO_3 .

3.2. NMR

In order to achieve "high resolution" in the solid state proton magnetic resonance spectra, we applied the flip-angle error and phase-error compensating version of the MREV-eight-pulse sequence^{1,13}. This sequence consists of $\pi/2$ -rf pulses with quadrature phases, applied at τ and 2τ intervals. The width of the $\pi/2$ -pulses was about $75 \mu\text{sec}$, τ was chosen as $4 \mu\text{sec}$ and the operating frequency of the spectrometer was 90 MHz. We point out that whereas homonuclear DD interactions are suppressed very efficiently with the MREV sequence, heteronuclear DD interactions are scaled down only by a factor of roughly 1/2. Multiple pulse single shot 1k-FT spectra were taken at 10° steps of azimuthal angle for each of the rotation axes a^* , b , and c for KHCO_3 , and at 5° steps for the rotation axes a , b , and c for KHSO_4 . TMS was used as an external reference, its signal recorded for each set of measurements and the "scaling factor" checked occasionally.

We measured T_1 of a powder sample of KHCO_3 at room temperature and obtained $T_1 = 5.8$ minutes, accordingly data for multiple pulse FT spectra were acquired every 5 minutes. Judging from the signal heights at different orientations of the crystals, the orientational dependence of T_1 appears to be small.

The proton T_1 in KHSO_4 is much longer, so data could be acquired only every $1\frac{1}{2}$ hours.

4. Results and Discussion

KHCO_3

Figure 3 shows three characteristic multiple pulse FT spectra of KHCO_3 , the upper two are from the b -crystal at azimuthal angles such that the single

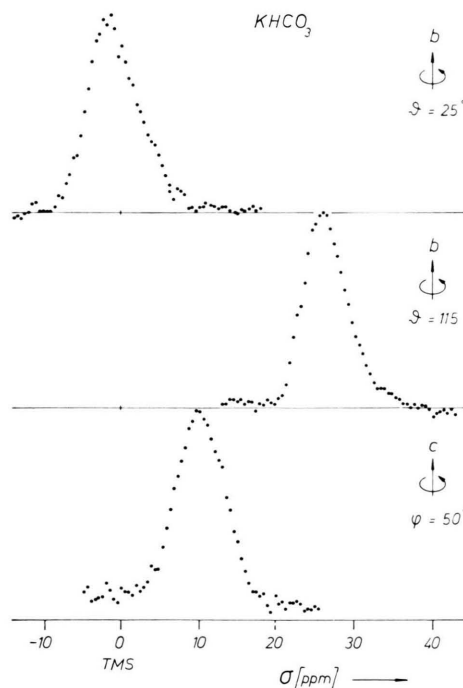


Fig. 3. Multiple pulse FT proton magnetic resonance spectra for three different orientations of single crystal of KHCO_3 .

line possible is at its extreme up- and down-field position; the bottom one is from the c -crystal in the orientation most favourable for a separation of the two possible lines. But neither in this nor in any other spectrum recorded is there any indication of a splitting of the two lines corresponding to the two types of dimers in KHCO_3 . Therefore, our data of the line positions allow us to determine only a single shielding tensor which is the average shielding tensor $\bar{\sigma}$ for the two types of protons (dimers) in KHCO_3 . Two of the principal axes of $\bar{\sigma}$, the X and Z axes, say, must lie in the a^* - c glide plane by crystal symmetry (see Figure 1). In this connection it is worth remarking that a deuteron study of

KDCO_3 ¹⁴ also failed to distinguish two magnetically inequivalent deuterons, although quadrupolar deuteron line splittings are much larger than chemical shifts of protons. The orientational dependence of the PMR line position is shown in Figure 4.

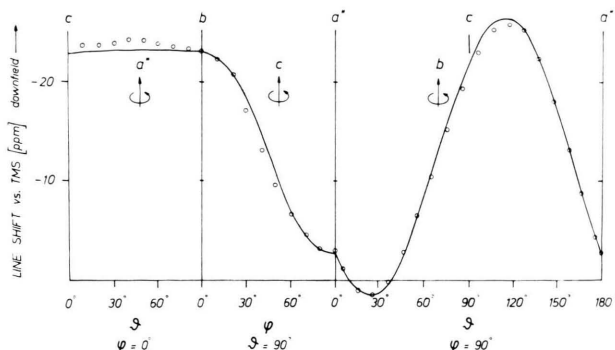


Fig. 4. Orientational dependence of PMR line position for KHCO_3 .

The curves are the results of a least squares computer fit. In all three rotation patterns the least shielded values are roughly the same, this means that \bar{Q} must be approximately axially symmetric¹. Moreover, the line position varies very slightly only for B_0 rotating in the b - c -plane, the unique axis of \bar{Q} must therefore be close to a^* and the principal values of \bar{Q} can be read off immediately from the patterns with b as rotation axis. The orientation of \bar{Q} is in good agreement with our expectations discussed in section 2: the maximum shielding is observed in "pattern b " at $\vartheta = 25^\circ$, this is just the orientation of the projections on the a^* - c glide plane of both of the hydrogen bond directions. The least shielded direction (at $\vartheta = 115^\circ$) corresponds to the normals of the dimers. These semi-quantitative results were refined by a least squares simultaneous computer fit of all of our spectral data of KHCO_3 . The results relative to TMS are:

$$\begin{aligned}\bar{Q}_{XX} &= -26.8 (\pm 1) \text{ ppm}, \\ \bar{Q}_{YY} &= -23.2 (\pm 1) \text{ ppm}, \\ \bar{Q}_{ZZ} &= +1.7 (\pm 1) \text{ ppm}.\end{aligned}$$

Two of the Euler angles (Φ and ψ , Goldstein convention¹⁵) specifying the orientation of the principal axes system of \bar{Q} relative to the a^* , b , c -frame are fixed by crystal symmetry to 90° , the third is found to be $\Theta = 68^\circ (\pm 2^\circ)$.

The reason why we failed to distinguish, as we had wished, the two magnetically inequivalent pro-

tons in KHCO_3 by two resolved NMR lines is two-fold: (i) the individual lines are strongly broadened by mechanisms to be discussed presently. (ii) as a result of the coplanarity of the dimers and of the smallness of the angle $\alpha = 4.7^\circ$ between the a^* - c -plane and the two different hydrogen bond directions, the orientations of the shielding tensors \bar{Q}_1 and \bar{Q}_2 of proton types 1 and 2 do not differ much which means that the shift of the two possible PMR lines with respect to each other remains small for all orientations of B_0 .

In order to assess how much of the observed line-width $\delta\nu_{\text{exp}}$, and of its orientation dependence (Fig. 5, upper curve), can be traced back to un-

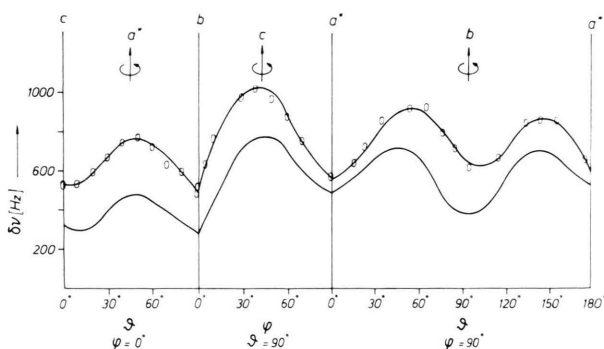


Fig. 5. Orientational dependence of PMR line width for KHCO_3 . Upper curves: experimental; lower curves: theoretical contribution to linewidth from ^{39}K - ^1H DD interactions.

resolved line splitting, we have to estimate the contributions of (at least) three line broadening mechanisms: (i) instrumental effects, (ii) residual ^1H - ^1H dipolar line broadening and (iii) line broadening by unaveraged heteronuclear ^1H - ^{39}K DD interactions.

From the width of lines in ^1H multiple pulse FT spectra of compounds without heteronuclei, the combined contribution to $\delta\nu_{\text{exp}}$ of (i) and (ii) can be estimated to be of the order of 120 Hz, and little angular variation is expected. In an attempt to get an estimate for (iii) we calculated the scaled contribution of the K^+ -ions to the second moment of the PMR line by the Van-Vleck formula¹⁶. We are well aware that the applicability of this formula to our case is somewhat doubtful because of the modifications of the ^{39}K -Zeeman levels by the ^{39}K quadrupole-electric field gradient interaction⁷. The strength of the latter is not known for KHCO_3 , but it may be assumed to be fairly small¹⁷ because

KHCO_3 is an ionic crystal. Therefore, we may reasonably hope that the Van-Vleck-formula yields at least qualitatively correct values for the linewidths. The results of this calculation are shown by the lower curves in Figure 5. The comparison with the experimental results shows that $\delta\nu_{\text{exp}}$ and in particular its orientational dependence are mainly due to ^1H - ^{39}K DD interactions. Most of the remaining difference is well accounted for by mechanisms (i) and (ii) and there is little left that can be ascribed to unresolved line splitting. This conclusion is supported by the observation that there is no systematic difference of $\delta\nu_{\text{exp}} - \delta\nu_{\text{calc}}$ for rotation pattern *b*, where unresolved line splitting is absent by crystal symmetry, and rotation patterns *a** and *c*, where a line splitting is possible in principle. This all is to the point that the angle β between the *a**-*c* plane and the unique axes of σ_1 and σ_2 must be small. In fact $\beta = \alpha (= \pm 4.7^\circ)$ is entirely consistent with our data and 9° is a conservative estimate for the upper limit of $|\beta|$.

KHSO_4

Figure 6 shows a typical multiple pulse FT spectrum of KHSO_4 with B_0 lying in the *a*-*b*-plane. Two spectral features are clearly resolved, but from symmetry considerations (Sect. 2) we expected four PMR lines. In contrast to the case of KHCO_3 , however, both resolved features show clear evidence for being composite lines. From such spectra the line positions and intensities can be determined with

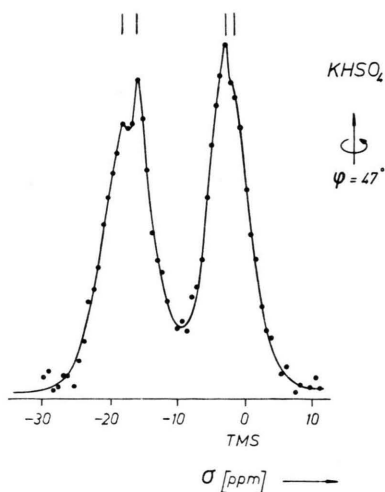


Fig. 6. Multiple pulse FT proton magnetic resonance spectrum of KHSO_4 . The lines at the top of the figure indicate guessed positions of component lines.

sufficient accuracy so that the rotation pattern for B_0 moving in the *a*-*b*-plane can be analysed in terms of four different lines corresponding to the four magnetically different protons in KHSO_4 . For B_0 moving in the *a*-*c* and *b*-*c* planes, at most two such distinct lines could be discerned. The data shown in Figs. 7 could be analysed by the least squares fit program mentioned above to yield the four shielding tensors $\sigma^{(a)} \dots \sigma^{(d)}$ shown in Table 1.

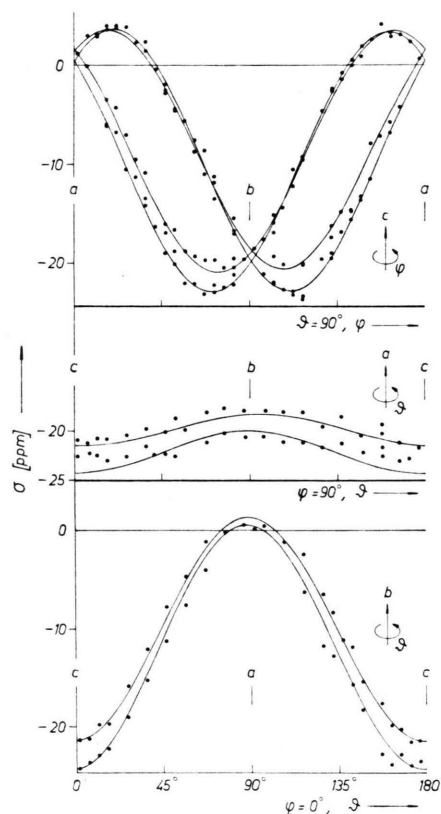


Fig. 7. Orientational dependence of PMR line positions for KHSO_4 .

Table 1. Proton shielding tensors in KHSO_4 . The principal components are given relative to liquid TMS. Φ , Θ , ψ are the Euler angles (Goldstein convention¹⁵) specifying the orientation of the shielding principal axes system to the crystallographic *a*, *b* and *c*-axes. See text for an assignment of sites *a*...*d* to particular protons.

site	a	b	c	d
σ_{xx} [ppm]	-21.6	-20.8	-24.3	-24.4
σ_{yy} [ppm]	-20.4	-20.1	-22.7	-22.8
σ_{zz} [ppm]	+3.5	+3.5	+3.7	+3.7
Φ	-72.7°	$+74.0^\circ$	-69.5°	$+69.4^\circ$
Θ	$+90.7^\circ$	$+89.3^\circ$	$+91.6^\circ$	$+91.8^\circ$
ψ	$+19.9^\circ$	$+20.6^\circ$	$+7.5^\circ$	$+12.3^\circ$

As discussed in Sect. 2 we expected — and obtained indeed — four distinct shielding tensors, but only two different sets of principal components. This is borne out by our results: $\sigma_{\alpha\alpha}^{(a)} \approx \sigma_{\alpha\alpha}^{(b)}$ and $\sigma_{\alpha\alpha}^{(c)} \approx \sigma_{\alpha\alpha}^{(d)}$ for $\alpha = X, Y, Z$. From this finding we conclude (i) $(\sigma^{(a)}, \sigma^{(b)})$ and $(\sigma^{(c)}, \sigma^{(d)})$ belong to crystallographically equivalent protons, respectively, and (ii) the differences $\sigma_{\alpha\alpha}^{(a)} - \sigma_{\alpha\alpha}^{(b)}$ and $\sigma_{\alpha\alpha}^{(c)} - \sigma_{\alpha\alpha}^{(d)}$ are a good measure for the experimental accuracy. The root of the mean square of these differences is 0.37 ppm.

On the basis of the data shown in Fig. 7, there is, unfortunately, no way of telling which of the pairs of shielding tensors should be assigned to the protons of the dimers and which to the protons of the chains. On the basis of the experience¹ that the most shielded direction for a proton in a hydrogen bond coincides closely with the direction of the bond itself, we propose, however, that $\sigma^{(a)}$ and $\sigma^{(b)}$ belong to the dimers while $\sigma^{(c)}$ and $\sigma^{(d)}$ belong to the chains.

In order to see that this assignment indeed gives the best agreement with the postulated parallelism of bond and most shielded direction, we first note that all the hydrogen bond directions as well as all the Z-axes of $\sigma^{(a)}$ through $\sigma^{(d)}$ fall closely into the a - b -plane ($\Theta^{(a)} \approx \dots \approx \Theta^{(d)} \approx 90^\circ$). The lines connecting the hydrogen bonded oxygens of the dimers make angles of $\pm 15.3^\circ$ with the a -axis. The corresponding numbers for the chains are $\pm 28.25^\circ$. The Z-axes of $\sigma^{(a)}$ and $\sigma^{(b)}$ make angles of $+(16.0 \pm 1)^\circ$ and $-(17.3 \pm 1)^\circ$ with the a -axis, respectively, and those of $\sigma^{(c)}$ and $\sigma^{(d)}$ $\pm (20.5 \pm 1)^\circ$. The close parallelism of the oxygen-oxygen line and the most shielded proton direction for the dimers points to a (dynamically?) symmetric hydrogen bond. The deviations — well outside of the experimental errors — between the O-O-lines and the most shielded H-directions for the chains points to the opposite for this type of hydrogen bond. The direction of the deviation is such that the angle between the most shielded H-direction and the S-O bond direction for one side of the hydrogen bond comes close to the tetrahedral angle. This is a hint to the point that the corresponding S-O-H bond angle is closer to the tetrahedral angle than it would be if the proton would occupy a symmetric position between the bonded oxygen atoms.

At first sight it may look disturbing that the tensors $\sigma^{(a)}$, $\sigma^{(b)}$ which we assigned to the dimers — apparently planar structures — should deviate very

little from axial symmetry, even less so than the tensors assigned to the chains. For planar structures the component of σ perpendicular to both the hydrogen bond and the plane is expected to be less by about 3–8 ppm than the component perpendicular to the bond, but in the plane, cf., e.g. KHCO_3 above and Reference¹. Directly above and below the “plane” of the dimer in KHSO_4 there are, however, $(\text{SO}_4)^{--}$ groups of neighbouring chains, and, using Gierke and Flygare’s theory⁸, it is easily shown that the Larmor-precession of the electronic charge of these groups adds a positive contribution to the out-of-plane component of $\sigma(\text{H})$, making it virtually equal to the in-plane-perpendicular-to-the-bond component.

Are the Hydrogens in KHCO_3 and KHSO_4 Predominantly Ionized or Neutral?

We have tried to use our shielding data and an analysis thereof following Gierke and Flygare (GF)⁸ to determine whether the hydrogens in the three types of hydrogen bonds encountered in KHCO_3 and KHSO_4 are predominantly ionized or rather neutral. This analysis closely follows our previous one for the hydrogen bond in solid KHF_2 and for details the reader is referred to Reference³. The procedure is as follows: Using the GF-theory we calculate $\sigma_{||} \approx \sigma_{zz}$ and $\Delta\sigma = \sigma_{zz} - 1/2(\sigma_{xx} + \sigma_{yy})$ with two adjustable parameters, R and ϵ . R is an empirical “reduction factor” introduced previously³ for term IV of the GF expansion of $\sigma^{(d)}$ in order to account for the fact that calculations of proton shielding tensors of hydrogen bonds consistently overestimate the shielding anisotropy $\Delta\sigma$. Its introduction may be justified on the grounds that the GF approach inherently assumes that the proton is well outside of the electronic charge distributions of neighbouring atoms or ions, an assumption which certainly is not fulfilled very well by the bonded oxygen atoms of a hydrogen bond. For this reason we apply R only to σ^{IV} (bonded oxygen atoms) which, however, dominates $\Delta\sigma$. The other parameter, ϵ , is a measure for the ionic character of the hydrogen bond, $\epsilon = 0$ means the hydrogen is fully ionized, $\epsilon = 1$ means the hydrogen carries one full electronic charge in a free-atom-1s-orbital. Term I of the GF-expansion of $\sigma^{(d)}$, i. e. σ_{iso} (free atom), is to be multiplied by ϵ . Neglecting as usual term III

of the GF-expansion, the equations for $\Delta\sigma$ and $\sigma_{||}$ read as follows:

$$\Delta\sigma_{\text{cal}} = R \cdot \Delta\sigma_{\text{b.o.}}^{\text{IV}} + \Delta\sigma_{\text{o.a.}}^{\text{IV}} + \Delta\sigma_{\perp}^{\text{V}} = \Delta\sigma_{\text{exp}}, \quad (1)$$

$$\sigma_{|| \text{ cal}} = R \cdot \sigma_{|| \text{ b.o.}}^{\text{IV}} + \sigma_{|| \text{ o.a.}}^{\text{IV}} + \varepsilon \cdot \sigma_{\text{iso}}^{(\text{d})}(\text{H-atom}) \stackrel{!}{=} \sigma_{|| \text{ exp}} \quad (2)$$

with $\sigma_{\text{b.o.}}^{\text{IV}} = \sigma^{\text{IV}}(\text{bonded O-atoms})$ and $\sigma_{\text{o.a.}}^{\text{IV}} = \sigma^{\text{IV}}(\text{other atoms})$.

The exclamation signs mean that R and ε should be adjusted to satisfy these equations. $\sigma_{\text{iso}}^{(\text{d})}(\text{H-atom}) = +17.8 \text{ ppm}^{18}$. Evaluation of $\sigma_{\text{b.o.}}^{\text{IV}}$ requires knowledge of the mean squared distance $\langle \varrho^2 \rangle$ of the oxygen's electrons from the oxygen nucleus*. In previous work on ionic systems^{3,19} we have taken weighted averages of $\langle \varrho^2 \rangle$ derived from Malli and Fraga's theoretical data²⁰ for the susceptibilities of O and O⁻. Comparing these data, which by the nature of the calculation are "free space" susceptibilities, with susceptibilities χ actually measured in solids²¹, we come now to the conclusion that Malli and Fraga's value for χ_{O^-} , which is equivalent to $\langle \varrho^2 \rangle_{\text{O}^-} \approx 1.7 \times \langle \varrho^2 \rangle_{\text{O}}$ is inappropriate for solids and that the value for the neutral oxygen atom, $\langle \varrho^2 \rangle_{\text{O}} = 3 \times 10^{-16} \text{ cm}^2$ is the best choice also for partly ionic systems such as KHCO₃ and KHSO₄. For the internuclear distances, also required for the evaluation of $\sigma_{\perp}^{\text{V}}$, we relied on the X-ray data^{10,11}.

As regards $\sigma_{\perp}^{\text{V}}$, the so-called spin-rotation term⁸ we know that its "parallel" component vanishes to the extent we can localize the origin of this term to the linear entity O...H...O²². $\sigma_{\perp}^{\text{V}}$, therefore, does not show up in Eq. (2) and $\Delta\sigma^{\text{V}} = -\sigma_{\perp}^{\text{V}}$. For $\sigma_{\perp}^{\text{V}}$ we use a former guess²³ of +16 ppm, the sign of which

is typical for linear structures²² and the size of which is at least consistent with the measured spin-rotation-tensor for the H₂O molecule²⁴. Table 2 summarizes the results of the calculations.

Table 2. Parameters used to determine, and results for R and ε .

	KHCO ₃	KHSO ₄ dimers	KHSO ₄ chains
$\Delta\sigma_{\text{exp}}$ [ppm]	27	24.2	26.8
$\sigma_{ \text{ exp}}$ [ppm]	33	34.4	34.5
$\Delta\sigma_{\text{b.o.}}^{\text{IV}} = 3/2 \sigma_{ \text{ b.o.}}^{\text{IV}}$ [ppm]	45.2	46.4	47.5
$\Delta\sigma_{\text{o.a.}}^{\text{IV}} = 3/2 \sigma_{ \text{ o.a.}}^{\text{IV}}$ [ppm]	7.2	0.7	6.8
R	0.79	0.85	0.76
ε	0.24	0.42	0.32

The main conclusion we draw is that in all three hydrogen bonds a large part of the hydrogen's charge has been drawn off by the bonded oxygen atoms ($\varepsilon = 0.24, 0.42$ and 0.32 for the dimers in KHCO₃ and KHLO₄, and for the chains in KHSO₄, respectively). This conclusion is not substantially altered even if our guess for $\sigma_{\perp}^{\text{V}}$ is in error by as much as $\pm 6 \text{ ppm}$. We also note that our numbers for ε compare favourably with the results of "population analyses" carried out by Ditchfield *et al.* for some model hydrogen bonds⁹. Reversing the flow of arguments, this can also be taken as a support of the simple GF approach, including the introduction of the empirical reduction factor R to be applied to $\sigma_{\text{b.o.}}^{\text{IV}}$, and of our guess of $\sigma_{\perp}^{\text{V}}$.

Acknowledgements

We would like to thank Professor K. H. Hausser for his interest in and support of this work. We are very grateful to Professor S. Haussühl for providing the KHCO₃ crystal. One of us (M.P.-S.) would like to thank the Max-Planck-Society for a stipend.

* Taking GF's definitions and notation, actually not $\langle \varrho^2 \rangle_n$ but $1/Z_n \langle \varrho^2 \rangle_n$ is the mean squared distance of the electrons from nucleus n , with $Z_n = \text{atomic number of nucleus } n$.

¹ U. Haeberlen, "High Resolution NMR in Solids: Selective Averaging", *Advances of Magn. Reson.*, Suppl. 1, Academic Press, London 1976, Review.

² M. Mehring, in "NMR: Basic Principles and Progress", Springer Verlag, Berlin 1976, in press, Review.

³ P. Van Hecke, H. W. Spieß, and U. Haeberlen, *J. Magn. Res.* **22**, 103 [1976].

⁴ T. K. Halstead, H. W. Spieß, and U. Haeberlen, *Mol. Phys.* **31**, 1569 [1976].

⁵ U. Burghoff, G. Scheler, and R. Müller, *Phys. Stat. Sol. (a)* **25**, K 31 [1974].

⁶ A. Pines, D. Ruben, S. Vega, and M. Mehring, *Phys. Rev. Lett.* **36**, 110 [1976].

⁷ H. W. Spieß, H. Zimmermann, and U. Haeberlen, *J. Magn. Res.*, in press.

⁸ T. D. Gierke and W. H. Flygare, *J. Amer. Chem. Soc.* **94**, 7277 [1972].

⁹ R. Ditchfield and R. McKinney, *Chem. Phys.* **13**, 187 [1976].

¹⁰ J. Thomas, R. Tellgren, and I. Olovsson, *Acta Cryst.* **B 30**, 1155 [1974]; **B 30**, 2540 [1974].

¹¹ D. Cruickshank, *Acta Cryst.* **17**, 682 [1964]; L. Loopstra and C. MacGillavry, *Acta Cryst.* **11**, 349 [1958].

¹² H. W. Spieß, H. Zimmermann, and U. Haeberlen, *Chem. Phys.* **12**, 123 [1976].

¹³ P. Mansfield and U. Haeberlen, *Z. Naturforsch.* **28 a**, 1081 [1973].

¹⁴ T. Chiba, *J. Chem. Phys.* **41**, 1352 [1964].

¹⁵ H. Goldstein, *Classical Mechanics*, Addison-Wesley, Reading 1959.

- ¹⁶ J. H. Van Vleck, *Phys. Rev.* **74**, 1168 [1948].
- ¹⁷ E. Lucken, *Nuclear Quadrupole Coupling Constants*, Academic Press, London 1969.
- ¹⁸ A. Carrington and A. D. McLachlan, *Introduction to Magnetic Resonance*, Harper and Row, London 1969, p. 55.
- ¹⁹ A. M. Achlama, U. Kohlschütter, and U. Haeberlen, *Chem. Phys.* **7**, 287 [1975].
- ²⁰ G. Malli and S. Fraga, *Theoret. Chim. Acta* **5**, 284 [1966].
- ²¹ W. R. Myers, *Rev. Mod. Phys.* **24**, 15 [1952].
- ²² C. H. Townes and A. L. Schawlow, *Microwave Spectroscopy*, McGraw-Hill, New York 1955.
- ²³ U. Haeberlen, U. Kohlschütter, J. Kempf, H. W. Spieß, and H. Zimmermann, *Chem. Phys.* **3**, 248 [1974].
- ²⁴ J. Verhoeven and A. Dymanus, *J. Chem. Phys.* **52**, 3222 [1970].